# TRUAX LANDFILL

## LEACHATE MIGRATION

STUDY

JANUARY, 1984

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Prepared by:

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Gentlemen:

Enclosed is a copy of a report drafted by the City of Madison Department of Engineering which compiles and discusses various leachate indicator parameters which have been detected in the groundwater monitoring wells near the Truax landfill site.

These wells are routinely monitored by the Engineering and Health Departments as part of the City's water quality maintenance and review program.

This report has been provided to Dane County, owner of the Truax site and Oscar Mayer Foods Corporation, which maintains private wells near the site. By copy of this letter, the report is also being provided to the owners of the former Burke sewerage treatment site.

If you have any questions regarding the report, please contact Jill Schmidt of the Health Department or David Benzschawel of my staff.

Very truly yours,

Mich.

Arnold E. Milke, City Engineer

AEM:DLB:map

cc: First Wisconsin National Bank c/o
Vernon J. Martin, Personal Representative
& Trustee of Henry E. Reynolds Estate
William Mundt, Murphy, Stolper, Brewster &
Desmond, Attorneys for Reynolds Estate
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# APPENDIX

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Water Supply Well Analysis

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#### I. INTRODUCTION

The Truax Landfill has been operated as an open-burning dump and sanitary landfill by two owners since the low area adjacent to the Starkweather Creek wetlands was first filled back in the 1930's. Commercial wastes were first disposed at the site in the late 1940's by the Oscar Mayer Company. It is assumed that a wide variety of wastes were disposed at the site by Oscar Mayer who apparently operated all or part of the site as an open-burning dump. Oscar Mayer probably disposed of both office and commercial wastes at the site. The City of Madison took over operations in 1953, and continued to use the area as a burning dump through 1960. Sanitary landfill operations began in 1961 and ended when the site was closed in 1972. Since the site was operational prior to the state legislation enacted in 1973 which required stricter environmental controls and long-term care, the State of Wisconsin Department of Natural Resources (DNR) has not required groundwater monitoring.

The landfill and the surrounding city-owned land was transferred to the County in 1973.

The Policy of the DNR for the "grandfathered" landfill sites which do not contain hazardous wastes is to reactivate groundwater monitoring and completely review the groundwater quality when signs of harmful groundwater contamination are evident. So at this time, the DNR does not require Dane County to perform any groundwater monitoring at Truax.

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Currently the City is monitoring Truax water quality on its own because the City Engineering Division and the City Health Department have agreed that the rising levels of leachate indicator parameters being detected in the groundwater monitoring wells are a cause for concern.

The City Engineering Division is involved at Truax because it is acting as engineering consul to the City Streets Division which is the licensee for all City landfills.

The City Health Department has an interest in the Truax Landfill because they are charged with protecting the environment of the City of Madison according to City Ordinance Sec. 7.46 and 7.47 which includes ground and surface waters.

## II. THE EXISTING GROUNDWATER TABLE

#### The Upper and Lower Aquifers

stone aquifer is being drawn down.

The United States Geological Survey prepared a report in cooperation with the City of Madison Water Utility on the upper and lower groundwater aquifers titled "Water Level Declines in the Madison Area, Dane County, Wisconsin." The report was drafted because of the concern for the dramatic drop in the lower aquifer levels due to municipal and industrial pumping in this aquifer by high capacity wells. The groundwater flow system under Madison is classified as a twoaquifer system composed of a confined lower sandstone aquifer underlain by a relatively impermeable bedrock and overlain by a leaky unconfined upper aquifer composed of glacial soils. Monitoring of the aquifers has shown that the upper aquifer is recharging the lower aquifer at an increasing rate as the lower sand-

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This change in traditional groundwater flow patterns has resulted in a reduction of the groundwater discharge to wetlands, lakes and streams.

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These findings particularly apply to the Truax-Oscar Mayer area as this area is situated adjacent to the wetlands of the east branch of Starkweather Creek and the area is surrounded by eight wells which pump about 8 million gallons of water per day from the lower sandstone aquifer.

Contours of the groundwater levels in the upper and lower aquifers are shown in Figures #1 and #2 in the Appendix. The figures show that one of the worst areas in the City for the channelization of flow from the upper to the lower aquifers is at the Truax Landfill area.

The Dane County Regional Planning Commission studied the Starkweather Creek drainage basin and determined groundwater pumpage at the highcapacity wells around Truax Landfill to be a major cause of the decline of flow in the creek. Figure No. 8 in the Appendix is a plan of the Dane County Regional Planning Commission's estimate of groundwater contours that have resulted from the heavy pumping of Oscar Mayer's wells.

## THE CITY OF MADISON'S CITY-WIDE DATA

The City of Madison's groundwater measurements confirm the draw-downs predicted by 1978 USGS report. Monitoring wells at Sycamore and Green Tree Landfills, both in areas of large groundwater depressions which have been monitored since 1973 to the present, show that the upper aquifer is being depleted by the local water wells.

The significance of this is that the leachate from Truax Landfill and the sludge beds will be drawn more quickly towards the adjacent water wells as the lower groundwater table drops.

The high capacity wells near Truax, installed by the City of Madison Water Utility and Oscar Mayer, have a cone of depression which extends about 2,600 feet from each well. Oscar Mayer has installed five wells on their property and it is estimated that the combined cone of depression from the wells would extend out from the wells about 3,500 feet on the average. These cones of depression are not uniform and are influenced by the permeability of the layered aquifers. The groundwater below the Truax Landfill appears to be strongly influenced by the cones of depression from the Oscar Mayer wells and the City of Madison Well #7 on Sherman Avenue at Schlimgen Avenue. These cones are shown on Figure #3 in the Appendix.

### SUMMARY OF GROUNDWATER FLOW STUDY

#### AT TRUAX LANDFILL, 1970

The "Hydrogeology of Solid Waste Sites in Madison, Wisconsin" included a study of the effect of Truax Landfill on the local groundwater system.

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It was written by R.J. Kaufman, a University of Wisconsin graduate student in geology as part of his doctoral requirements. His data concerning the flow patterns between the upper and lower aquifers is in agreement with the assumptions used by the USGS five years later when they computer-modeled the groundwater flow patterns in Dane County. Kaufman's data from the first year of the study showed that the upper groundwater flow pattern was westward under the landfill and that the vertical gradients were downward for most of the area, including the area immediately adjacent to Starkweather Creek and its ditches. This included the undisturbed marsh areas and the man-made sewage sludge irrigation fields. The shallow groundwater flow patterns are shown in Figure #4 of the Appendix.

Kaufman's data shows that the pumping in the adjacent high-capacity water wells was already drawing in groundwater from the upper unconfined aquifer. In his report, he speculates that "the change (from an upward to a downward flow) is at least partly due to groundwater withdrawal via nearby high-capacity wells, . . municipal Well #7 and the Oscar Mayer Company wells." His water level readings in 1968 and 1969 show a constant westward flow direction for the majority of the landfill and the Burke plant sludge lagoons, #1 thru #7. The cause for this flow pattern was the cone of depression he measured along the west side of the landfill. These patterns are shown in Figure #3 in the Appendix.

Kaufman decided that the severe drop or trough in the groundwater table was part of the cone of depression from City Well #7 and Oscar Mayer wells. Kaufman reviewed the data and concluded "Reversal of the gradient so as to cause lateral flow out of the marsh is apparently a result of a combination of permeable drift, shallow depth to rock and nearby high capacity

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wells. Development of the pumping cone has proceeded eastward under the landfill, thereby inducing flow generally to the west and northwest. The remaining water in the landfill, primarily along the southern border and, to a lesser extent, along the eastern edge is not affected by the diversion and as a result is discharged locally to the ditch system and the marsh. Drainage of landfill leachate eastward toward the marsh is indicated only on the maps prepared from data collected in September and November, 1969, and may represent the exception. In any event, the percentage of area draining toward the marsh and the very slight gradient indicate small flow volumes compared to the total discharge from the landfill and the sewage disposal area.

Relative permeabilities of the glacial sediments also are related to the flow pattern in the landfill and sewage disposal areas. The cone or trough is most developed in the sandy morainal material below and adjacent to the northwestern quarter of the landfill. Drainage or diversion is easiest because of relatively high permeability compared to the more clayey materials present in the southern and southeastern parts of the landfill. Thus the cone has expanded along a path of least resistance until the clayey materials are reached. The marsh sediments have a low permeability and, in effect, constitute a boundary condition and an approximate limit on further eastward expansion of the cone."

It appears that the local clay and clayey silt deposits on the top of bedrock are a key to the groundwater flow patterns in the upper unconsolidated aquifer. A review of the bedrock and clay locations, shows that the most influential clay deposit is a roughly circular deposit varying in depth from 30' to 120' which lays on top of the upper weathered sandstone bedrock just south of the Burke Sewage Treatment Plant.

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This clay deposit which has a very low permeability, blocks the cone of depression from the surrounding wells from reaching the groundwater in the upper sand aquifer located above it. This clay boundary along with the high recharge from the sludge lagoon area, creates the groundwater mound under the sludge lagoons. This mound is shown in Kaufman's data used to draw Figure #5.

# CITY OF MADISON'S TRUAX DATA

The City's monitoring from 1970 to present confirms that the flow patterns mapped by Kaufman still remain the same. Figure #6 is an eastwest cross section of the landfill showing geology and flow patterns. Figure #3 shows the cones of depression and groundwater contours. The glacially deposited soils in the vicinity of Truax Landfill have been studied to determine their boundaries and soil types. Using the average porosities and permeabilities of these types of soil, it is anticipated that the actual horizontal flow velocities in the medium-sized silty sand aquifers will vary between 40' per year to 120' per year as the groundwater moved one-half mile towards the water supply wells.

Groundwater velocities of 80' per year match fairly well with the measured leachate contours at various monitoring wells down gradient of the landfill. The Engineering Division has been monitoring five wells along the west side of the landfill. Representative water quality data showing the movement of the leachate plume is shown in the attached Table I.

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# TABLE 1

GROUNDWATER	ANALYSIS
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	Well No.	<u>4/77</u>	5/78	11/79	<u>5/80</u>	<u>3/81</u>	<u>4/82</u>	7/82	<u>4/83</u>	<u>10/83</u>	2/84	Well No.
Chloride	101	-	-	-	-	63	68.5	80	80	92	80	101
(Mg/L)	104	DRY	DRY	87	<u>-</u>	27	18	40	35	24	17	104
	121-A	3,900	Damaged	Casing-Dama	ged Casing	5,130-		5,650	5,350	5,500	5,400	121-A
	200D	-	-	-	-	-	66	30	46	32	41	200D
	200S	-	-	-	-	-	0	1	225	24	62	200S
	152	-	-	-	52	137	-	Construction	in Area	-	180	152
	OSC-5	49	42	-	-	42	-	-	-	. –	48	0SC-5
Specific	101	-	-	-	-	1,610	1,490	1,580	2,210	2,200	2,100	101
Conductance	104	DRY	DRY	2,100	1,930	2,100	1,820	2,200	2,220	2,200	1,820	104
(MMHOS/CM)	121-A	16,800	Damaged	Casing-Dama	ged Casing	15,000-		15,000	17,100	16,600	15,700	121-A
	<b>20</b> 0D	-	-	-		-	1,010	630	880	700	860	200D
	200S	-	-	-	-	-	320	380	2,200	560	960	200S
	152	-	-	-	600	1,960	-	Construction	in Area	-	1,150	152 -
	0SC~5	825	810	-	-	870	-	-	-	-	820	0SC-5
Sodium	101	-	-	-	-		96	84	100	90	84	101
(Mg/L)	104	DRY	DRY	22	18	23.5	42	23	43	30	20	104
(	121-A	1,470	Damaged	Casing-Dama	ged Casing	1,570-		1,965	1,950	1,934	1,870	121-A
	200D	-	-	-	-	-	41	21	40	32	29	200 - D
	200S	-	-	-	-	-	29	14	250	41	90	200-S
	152	-	-	-	27	31.5	-	Construction	in Area	-	42	152
	OSC-5	18	18	-	-	19	-	-	-	-	22	0SC-5
Calcium	101	-	<del>_</del> `	-	-		124	112	153	142	199	101
(Mg/L)	104	DRY	DRY	-	202	213	216	225	228	231	231	104
(8, -)	121-A	630+	Damaged	Casing-Dama;	ged Casing	888-		900	948	840	788	121-A
•	200-D	-	-	-	-	-	98	48	69	·· 62	75	200-D
	200-S	-	-	-	-	-	38	40	142	57 -	80	200-S
	1.52	-	-	-	12	195	-	Construction	in Area	-	15	152
	OSC-5	88	75	-	-	78	-	-	-	-	84	OSC-5
Magnesium	101	-	-	-	i 🕳		87	90	132	136	50	101
(Hg/L)	104	DRY	DRY	-	128	132	111	126	135	130	138	104
,	121-A	138+	Damaged	Casing-Dama;	ged Casing	212-		222	240	240	. • 54	121-A
	200-D	-	-	-	-	<b>-</b> `	66	33	48	42	52	200-D
	200-S	-	_ <b>-</b> -	-	-	-	11	16	29	46	21	200-S
	152	-	-	-	46	106	-	Construction	in Area	-	99	152
	OSC-5	49	48	-	-	45	-	-	-		45	OSC-5

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#### III. GROUNDWATER QUALITY

#### Truax Landfill Leachate

All groundwater data from 1968 to the present confirms the presence of two groundwater plumes. The Truax Landfill has created one plume and the Burke Sewage Treatment Plant has created another. Due to the size of the treatment plant, its plume could be subdivided into the north irrigation field plume and the south irrigation field/ sludge lagoon plume.

Since the north irrigation field lacks sufficient monitoring wells and any operable wells have not been monitored since 1968, it is difficult to analyze what effect it has on the adjacent landfill plume. It would be safe to assume that these two groundwater flows are mixed along their common boundaries. It appears however, that the leachate being detected west of the landfill is coming mainly from the landfill with minor contamination from the irrigation field.

Figure #3 and #6 show the estimated limits of the leachate plumes. Truax Landfill was operated as an open-burning dump from the late 1940's until 1960. It is assumed that most of this waste was placed directly on the existing native soils at the low end of the landfill in the southeast corner bounded on both sides by drainage ditches. No liners of clay were placed under any of the waste and no leachate collection system exists at Truax. Almost all leachate percolates downward until it reaches the groundwater table.

Records show that when the City of Madison operated the site as an open-burning dump, almost all of the waste hauled to the site was combustible.

From 1961 until 1972, the site was operated as a sanitary landfill and it received almost all residential waste.

A groundwater contour was drawn on Figure #3 for a chloride ion concentration of 80 Mg/L and a specific conductance of 800 UMHOS based on monitored levels in the groundwater wells. This contour coincides with a groundwater plume which would have an average velocity of 80' per year which is a typical velocity for the soils encountered at Truax. The limits of the landfill leachate plume are shown on Figure #3. The groundwater monitoring shows that the landfill leachate is migrating off of the landfill site and has probably reached Oscar Mayer Well #5 and will reach the City Well #7 within 10-20 years.

## BURKE TREATMENT PLANT LEACHATE

As mentioned previously, the leachate leaving the plant can be divided into two plumes. The one closest to the water supply wells is the one of greatest concern. This plume was fed by the south irrigation field and the sludge lagoons.

The plant was operated from 1914 to 1926 using a contact bed and trickling filters. It was closed down in 1936 and then used by the US Army from 1942 to 1945 for treating wastes from Truax Air Field.

The Oscar Mayer Company reactivated the plant in 1951 to treat meat packing wastes prior to discharging them to Madison Metropolitan's Treatment Plant. Sludge was lagooned on the site on a year round basis. The Burke Plant was last operated in 1976.

The monitoring well network along the west and south side of the treatment plant area is inadequate to detect the exact location of the treatment plant leachate plume.

But, based on the average groundwater velocities for the underlying sandy soils and the groundwater velocity detected under the landfill, the treatment plant leachate plume has had more than enough time to reach the Oscar Mayer supply wells.

The clay deposits directly below the treatment plant have slowed the leachate plume but have not stopped its migration towards the wells. It is expected that the leachate is being drawn around both sides of the roughly-circular clay deposit. Groundwater troughs exist on both sides of the clay barrier and the trough along the west side has monitoring wells which have detected a sewage leachate plume.

#### MONITORING WELL DATA

From Figure #3 it can be seen that the leachate plumes from the two sources are mixing along a variable boundary which passes near Wells #152, 101 and 200.

Well #101 is 35' deep and too shallow to pick up the landfill leachate plume, although it has detected the top of the sludge lagoon plume at various times.

Well #200 may have been contaminated by both plumes, but the evidence is not conclusive.

Well #152 it appears, is contaminated with landfill leachate only. The well water analyses are shown in Table #1 of the Appendix. The relative strengths of the two plumes are shown in Table #2. The strengths of the various parameters at a point in the leachate plume 1,000 feet down gradient of the source were estimated from all monitoring well data on the site and from published data on similar treatment plants and landfills. (See Attached Table #2)

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# TABLE #2

# ESTIMATED LEACHATE STRENGTHS 1000' DOWN GRADIENT

		Typical Sewage Treatment Plant
Parameter	Typical Landfill Leachate	Leachate
Chloride	120	200
Spec. Conductance	2,000	2,000
Sodium	40	100
Calcium	60	100
Magnesium	40	40
COD	80	150
Phosphorus - Total	.10	2.0
Potassium	5	20
BOD	8	40
Organic Nitrogen (Ammonia Indicator)	0.5	3.0

This data confirms the conclusion that separate leachate plumes are migrating towards the water supply wells.

The table shows that the sodium, potassium, ammonia, phosphate and BOD levels can be used to determine the source of the leachate. Kaufman discussed using these same parameters to determine leachate plume sources in his study in 1970. He felt that the levels of sodium, potassium, nitrate plus nitrite, ammonia, phosphorus and chloride would all be high in ground water known to be polluted by sewage. His report also included an analysis of the sanitary wastes treated by Oscar Mayer Company which is shown in Table #3.

Of particular importance, are the exceedingly high concentrations of nitrogen and phosphorus and the high pH level. The most polluted ground water sampled for the 1970 study was under the sludge lagoons and had an ammonia level of 314 Mg/L and a total phosphorus level of 8.91 Mg/L. (See Table #3 Attached)

## TABLE #3

# QUALITY OF SANITARY WASTES FROM THE OSCAR MAYER COMPANY $^1$

Sampl	le 1	Loc	at	ion

Parameter	Pretreatment Plant Sludge	Burke Plant 	Burke Plant Digester Sludge	Internatant From Lagoons	Secondary Effluent
pH	6.5	6.5	7.5	7.5	7.88
Specific Conductanc	e 3100	2800	9100	7100	
Soluble Na mg/l	200	500	700	300	370
Soluble K mg/l	50	20	140 <sup>4</sup>	60 <sup>5</sup>	18
Soluble Mg mg/1	100	100	1004	90 <sup>5</sup>	74
Soluble Ca mg/l	400	100	300 <sup>4</sup>	130 <sup>5</sup>	29
NO3-N	30	50	50	1	
NH3-N	192	96	840	5800	40 <sup>2</sup>
_ganic N	1600	1600	1000	900	15 <sup>2</sup>
Total P	640	480	310	83	
Ortho P	11	20	17	21	
C1	50	550	5504	325 <sup>5</sup>	770
COD	116000	65900	22500	1700	
Total Soluble P	77	93	53	29	10
BOD	40000 <sup>3</sup>	)000-30000 <sup>3</sup>			138
Suspended Solids	20	1000-20000-			90

- 1. All samples analyzed by W. Noel, City of Madison Health Department. Twenty secondary effluent samples collected from December 1966 to June 1967; mean values shown. Other wastes sampled and analyzed in May 1970; value shown based on one analysis of each waste category.
- 2. Concentration in effluent irrigated to North and South Fields.
- 3. D. Dencker, personal communication.

4.  $\pm$  50 mg/l accurace due to problems with dilution and color

5.  $\pm$  20 mg/l accuracy due to problems with dilution and color

Another comparison between landfill leachate and domestic waste waters was made by D.R. Brunner and R.A. Carnes and was used in a paper they authored which compared the relative strengths of leachates. Their data was used by the U.S.E.P.A. in one of the EPA solid waste manuals on groundwater monitoring. The Table is indicated below:

## TABLE #4

## CHARACTERISTICS OF LEACHATE AND DOMESTIC WASTE WATERS

Constituent	<u>Old Leachate</u>	Waste Water
Iron (Fe)	1.5	0.1
Magnesium (Mg)	81	30
Potassium (K)		
Sodium (Na)		
Phosphate (P)	4.96	10
Total N	7.51	40
BOD5		200-
COD	81	500

This table substantiates the preceeding tables and City water quality data gathered at Truax. The landfill plume should have a higher iron, magnesium, calcium and a pH around 6.0 and 7.0. The waste water plume should have a higher potassium, sodium, phosphate, nitrogen, BOD, COD and a pH around 7.0 to 8.0.

Wells #152 and #200 which are located on the boundary between the two leachate plumes, are beginning to detect groundwater with parameters that are approaching the levels expected for the landfill and waste water plant leachates.

The water analyses for Truax Wells #200 shallow on April 20, 1983, and for #152 on July 23, 1981 as follows:

PARAMETER #:	200 (Mg/L)	#152 (Mg/L)
COD	183	251*
Organic Nitrogen	6.5	
Chlorides	225	168
Conductance (UMHOS/CM)	2200	2160
Potassium	25.4	2.5
Sodium	250	33.5
Calcium	142	166
Magnesium	29	109
рН	7.5	6.96
Iron	.13	4.37
Phosphorus (Ortho)		.12
BOD	39.6	6.7

\*Data suspect: Normal Range = 25.0

A cross section, Figure #6, was drawn along the center line of the landfill leachate plume based on groundwater monitoring at Truax and data gathered at landfills located in similar soils with approximately the same groundwater velocities.

The monitoring well network at Truax is inadequate to draw firm conclusions on the exact location of the plume, but the data generated by Wells #104, 152, 200 and Oscar #5 was adequate to construct Figure #6. It should be pointed out that Oscar Mayer Well #5 is already experiencing a slight drop in water quality.

It appears that either of the two leachate plumes could be the cause of the water degredation. The water analysis for Oscar Mayer Wells #2 and #5 and City Water Utility Wells #7 and #5 are shown in the Appendix. Oscar Mayer Well #5 is 35 Mg/L higher in chlorides, 200 UMHOS/CM higher in conductance, and significantly higher in sodium, calcium, and magnesium than all of the surrounding high-capacity water supply wells.

## IV. PROPOSED MONITORING PROGRAM

# Leachate Parameters of Concern

Groundwater flow patterns based on the water table gradients and soil permeabilities are shown in Figure #3. It is assumed with reasonable certainty that the leachate plumes from the Truax Landfill and Burke Sewage Treatment Plant have already reached the bedrock aquifer at Oscar Mayer Well #5.

It is not certain which plume has migrated through the full depth of the upper sand aquifer and reached the lower sandstone aquifer in which the water supply wells are screened, but the Burke plume had a 20-30 year head start on the landfill leachate plume, and is more suspect. One or both plumes could be the cause of groundwater deterioration at Oscar's #5.

The groundwater flow patterns also show that the landfill plume is getting close to City Well #7. At this point, no contamination has been detected, but the leachate will invariably reach the well.

It is anticipated that the plume would reach #7 sometime within the next 20 years. The time it takes for the plume to reach #7 is dependent upon the well's pumping rate and how much the flow in the sandstone aquifer is channelized.

The leachate parameters that will be of concern as the wells become more contaminated are dependent upon the Federal Drinking Water Quality Standards. The parameters of concern cannot be selected at this time because of a lack of information from the existing groundwater monitoring network at Truax.

The City of Madison needs more information concerning the levels of heavy metals, organics and some other parameters which have been detected in the leachate but not reliably. The parameters already detected in the leachate which should be closely studied because of their potential for exceeding the drinking water standards are: nitrates, organic contaminants, manganese, cadmium, sodium and chlorides.

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#### BURKE MONITORING PROGRAM

The City Engineering Division has no responsibility for the long-term care of the Burke Sewage Treatment Plant, but the City Health Department has been assigned the responsibility for protecting the environment which includes groundwater.

It is questionable whether the City Health Department should spend money studying the leachate plume from the Burke Treatment Plant since it is owned by someone else, it was operated for a considerable amount of time by Oscar Mayer Company and it is probable that the only well to be contaminated by the plume is Oscar Mayer Well #5.

The groundwater monitoring network around the Burke plant appears to be completely inadequate. Almost all of the 20 or so wells around the plant have screens which are set too shallow to detect any of the leachate. It would be very expensive to redrill all new deep wells that would be required to adequately monitor the Burke plant. The well construction alone could cost as much as \$30,000. At this time, it appears the City of Madison should leave the treatment plant leachate plume alone and concentrate on getting the County to start a expanded monitoring program for the landfill. It is probable that the Department of Natural Resources will require the current owner of the Burke Treatment Plant property to start a monitoring program for that site.

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#### TRUAX MONITORING PROGRAM

Since the Truax Landfill is now owned by the County and the monitoring program is to be expanded, the City should start negotiating with the County to accept the responsibility for the groundwater monitoring program.

As mentioned previously, there currently exists a lack of data for certain groundwater parameters which may be contaminating the groundwater. The levels of organic contaminants, which have not been monitored at all, and the levels of heavy metal which have been measured sporadically should be monitored on a regular basis.

Since leachate has been positively detected, the monitoring schedule should be set so that any significant change in the strength or rate of leachate movement is detected.

An effective monitoring program can be sustained if Wells #152, 200 deep, 132, 200 shallow, 121 deep, 121 shallow, 104, 101, City Water Well #7, and Oscar Mayer Well #5 are monitored for the following parameters:

Quarterly --

5-Day BOD COD Phosphorus Ammonia Nitrogen Nitrate pН Alkalinity Hardness Chlorides Specific conductance Soluble potassium Soluble sodium Soluble calcium Soluble Magnesium Sulfate Silver Barium Cadmium Chromium Cobalt Iron Lead Manganese Nickel Zinc

-21-

Semi-Annually --

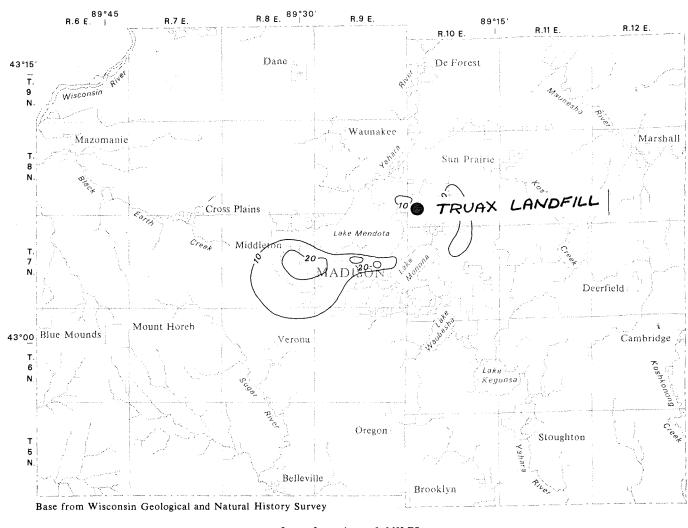
Organic Contaminants (to be selected) Selenium Arsenic Mercury

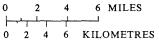
Well #152 is a key well. It is the only well located towards the center of the leachate plume. Unfortunately, the well's screen is located only 5' below the groundwater surface. Should the data from the well become irregular or the groundwater table drop, the well should be replaced with a well nest. The new nest should have two wells with screens set approximately 65' and 85' deep.

If Well #152 remains functional, then no additional wells will probably be needed. Should the existing wells show that the levels of contaminants are increasing more quickly than anticipated and the drinking water standards are in danger of being exceeded, then a new well nest should be located down gradient of Well #152. Any new well should be located so that the attenuative ability of the soils can be estimated. If new wells are drilled, soil samples should be taken so that groundwater velocities can be determined from permeability tests.

### BIBLIOGRAPHY

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- Kaufman, R.F. 1970. Hydrogeology of solid waste disposal sites in Madison, Wisconsin. Ph.D. Thesis. Univ. Wisconsin, Madison
- McCleod, R. S. 1975. A digital computer model for estimating hydrologic changes in the aquifer system in Dane County, Wisconsin. Wis. Geol. and Nat. History Survey Information Circular No. 30. Madison, WI 40 p.
- McCleod, R. S. 1978. Water-level declines in the Madison area, Dane County, Wisconsin. USGS Open-File Report 78-936. Madison, WI 15 p.
- U.S. Environmental Protection Agency. 1980. Procedures Manual for Groundwater Monitoring at Solid Waste Disposal Sites. EPA: SW-611 Washington, DC 269 p.

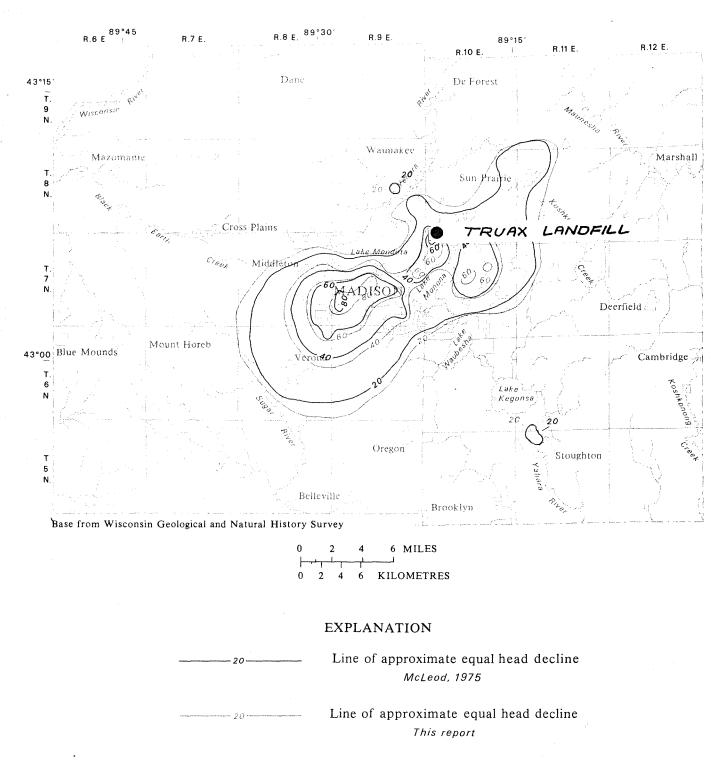




# **EXPLANATION**

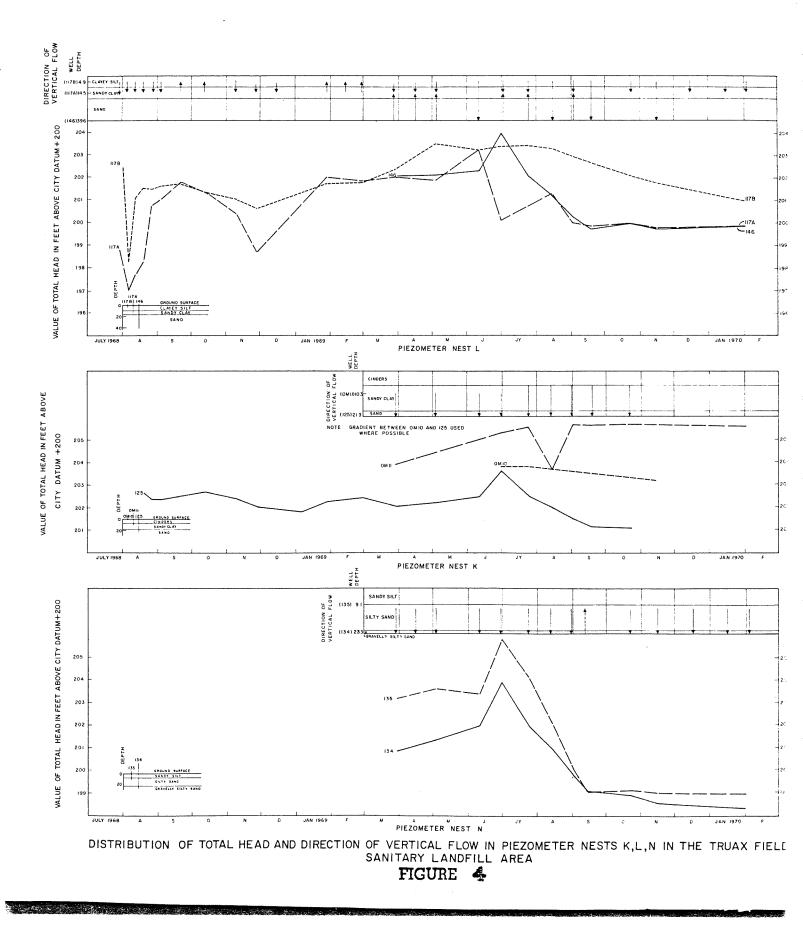
Line of approximate equal head decline Interval 10 feet (3 metres)

Figure **1**. Computed head declines in the upper aquifer by 1990.

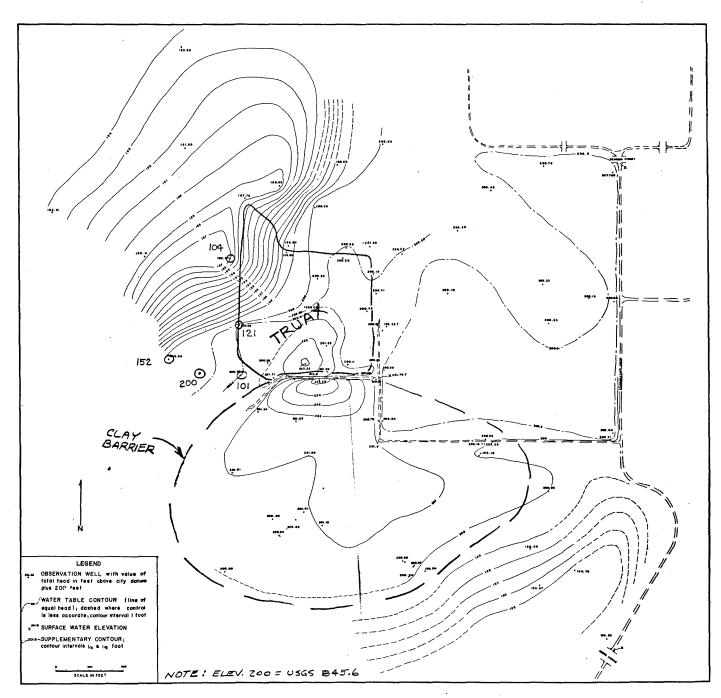


Interval 20 feet (6 metres)

Figure 2 Comparison of computed head declines for the sandstone aquifer, 1990.



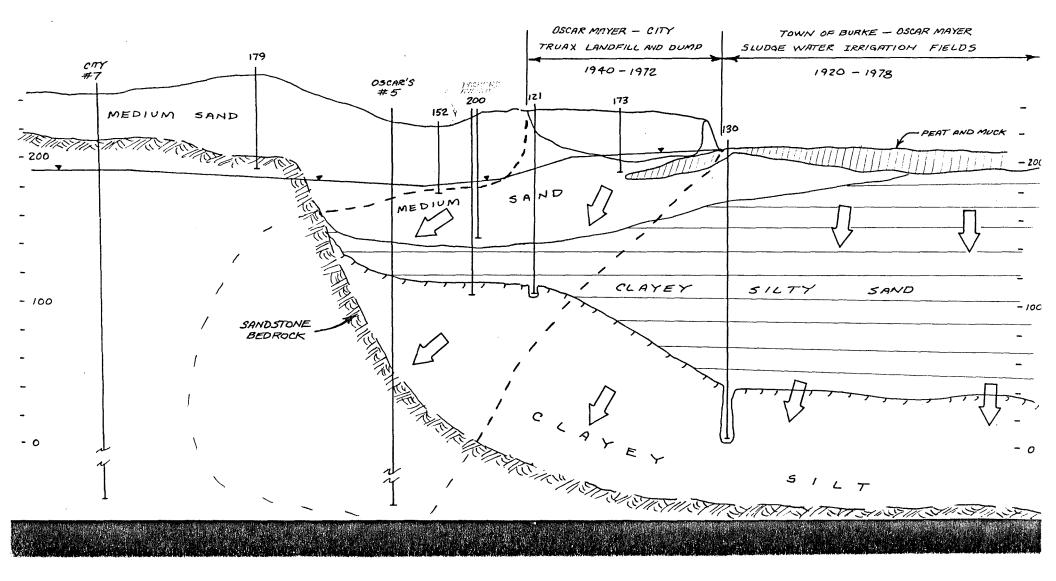
FIQURE 5

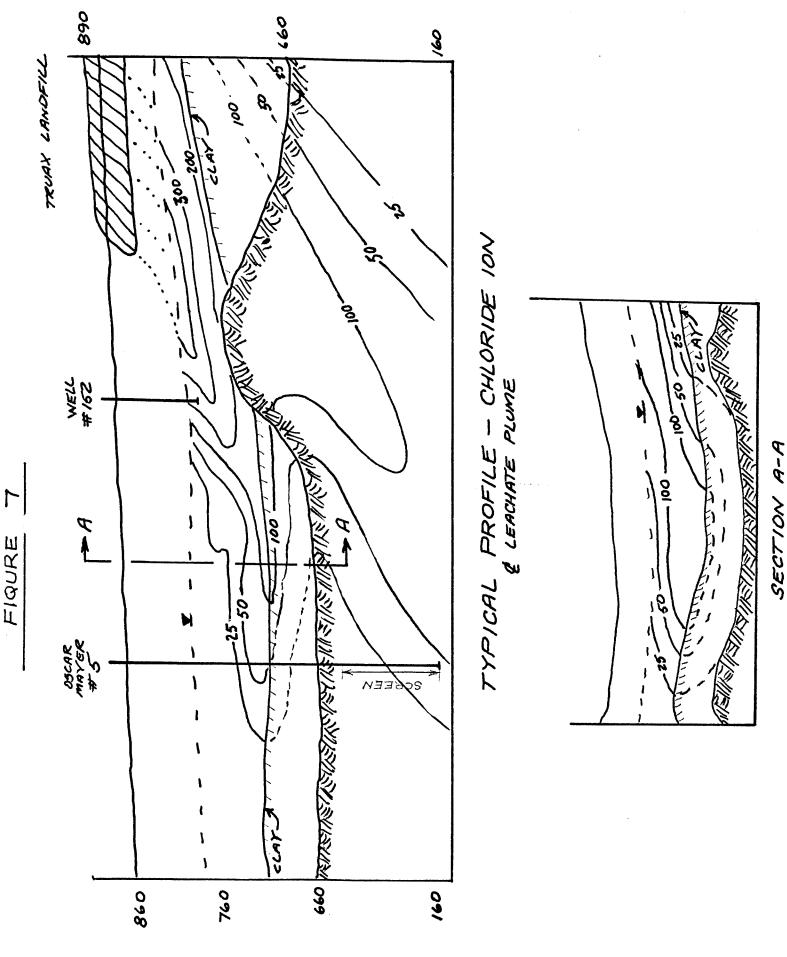


MAP OF THE TRUAX FIELD SANITARY LANDFILL AREA, SHOWING CONFIGURATION OF THE WATER TABLE, FEBRUARY 5, 1970

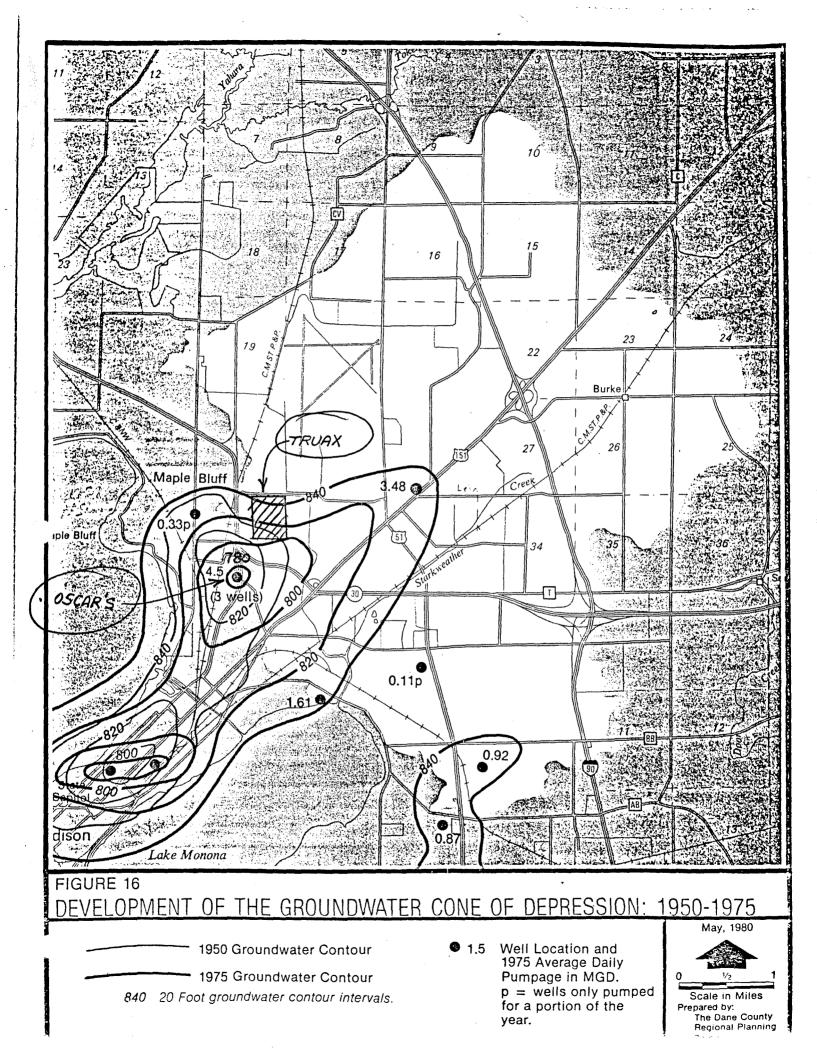
FIQURE 6

11、11月14日11月14日(14月14月14月)(14月14日)(14月14日))(14月14日))(14月14日))(14月14日))(14月14日))(14月14日))(14月14日))(14月14日))





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# Water Analysis Report

	PHYSICAL PROPERTIES		_			_F	Reported
$\mathbf{Y}$	1pH at 25°C 7.6		_		mg/l		
		/30 H2SO4		CS in Water		Plan	t Name
K		/30 H <sub>2</sub> SO	<b>_</b>	CS in Glycol			ess
		/30 H <sub>2</sub> SO4	1	Na_EDTA-Total			
		/44 NaOH		Na <sub>3</sub> NTA-Total		Atte	ntion
		/44 NaOH		Na <sub>3</sub> NTA-Free			of Samp
-		mhos/cm		Na <sub>4</sub> EDTA/NTA-Total		1	ole Point
		mhos/cm		Na,EDTA/NTA-Free			
-	Dissolved Solids @ 180°C 465	mg/l	1			Date	ofColle
	Suspended Solids @ 105°C < 5	mg/i				P	LANT
1	Total Solids @ 180°C 475	mg/I				: RE	SULTS
	Specific Gravity			METALS*		-	
-	**************************************	PHA units			mg/l	7:	н
1	Turbidity	J.T.U.		Jron (Fe) <b>2</b> 0,05		11	
-	Sample Description (visual): water white	Clear 1	Ľ	Copper (Cu)			10
			4	Manganese (Mn) 20.05		E	<u></u>
	CHEMICAL ANALYSIS			Aluminum (Al)			Acidity
1		mg/l		Nickel (Ni)			eutralize
	Hydroxide (OH)			Zinc (Zn)		1-15	onductiv In-neutra Inductiv
	Carbonate (CO <sub>3</sub> )			Calcium (Ca)			onductiv Dissolved
	Bicarbonate (HCO <sub>3</sub> ) $4/2$			Magnesium (Mg)			iolids Susp. Sol
	Carbon Dioxide (CO <sub>2</sub> )			Hex Chromium as CrO₄-dissolved			otal Soli
$\leq$	Ammonia (NH <sub>3</sub> ) <0.05			Total Chromium (CrO₄)			
1	Chloride (CI) 49			Tri Chromium as CrO <sub>4</sub> -calc		·	Color
Le	-Sulfate (SO <sub>4</sub> )	•				·	furbidity
1	Silica (SiO <sub>2</sub> ) / 5			·			CO <sub>2</sub>
$\overline{\mathcal{L}}$	Nitrate (NO <sub>3</sub> ) $< 5$					;	
1	Ortho Phosphate (PO4) <0,/			- TRACE ANALYSIS		1	dent
14	Polyphosphate (PO4)-calc. 0,2				mg/l	C	Chemical
4	$\partial$ rganic Phosphonate (PO <sub>4</sub> )-calc. < $0, 1$			Trace Chloride			Spec only
				Trace Silica		<b>P</b>	
				Trace Iron		Eng	ineer A
				Trace Copper		521-	Internati
1	Hardness as $CaCo_3$ -dissolved $420$			Trace Nickel		529-	Special A
1	Calcium (Ca)-dissolved 88 .			Trace Zinc		, <del></del>	Wholesal
V	Magnesium (Mg)-dissolved 49			Trace Calcium (1) (2) (3)		1	
./	Sodium (Na)			Trace Magnesium (1) (2) (3)		نذ مسمعتنام	فليسكنك ويستهيه تعرره بالامو
	Potassium (K)			Trace Sodium (1) (2) (3)			
	· · · · · · · · · · · · · · · · · · ·					Sod	ium (Na)
	Total Organic Carbon (TOC)					Sulf	ur (S)
	Colloidal (Non-reactive) SiO <sub>2</sub> -					Vana	adium (V)
	Şolveht (freon) Soluble			. 9.39/9.17		Ash	
	Surfactant-MBAS			<u></u>		API	Gravity

HTTLOFF JJU67 Reviewed by: Reported: ant Nam ddress ttention pe of Sample wo 0 ample Poin ate of Collection Time PLANT RESULTS PROCESS WATER ANALYSIS Pe NHS Trace Fe TOH A er Cu Trace Cu MO so. + Mn Trace CI SIÓ Zn Trace Na 8 AI NÓ Trace SiO Acidity Neutralized Conductivity Un-neutralized Conductivity Dissolved Solids Conordan Ni PO4-ortho cs PO4=poly CrO4-Total PO\_Organic CrO\_-Tri Syndet Na4EDTA-T Susp. Solids Hardness CrO₄-Hex Na3 NTA -free Car Total Solids Color Ńд Na Turbidity TOC Solvent Sol. CO2 DEPOSIT ANALYSIS Ident 1 Chemical Spec only FOR CALGON USE ONLY Oldan Office Code ngineer 1 525- Cooling and Municipal 21-International 523-Oilfield 531-Boiler Chemicals 532-Hall-Bradford 29-Special Accounts 90-Wholesaler Handling-FORM NO. 5060M  $\sim 4$   $\pm$ · C; B FUEL OIL ANALYSIS odium (Na) ppm ulfur (S) % anadium (V) ppm sh %

the transford otherwise, all metals are total (dissolved & particulate)



**Calgon Analytical Laboratories** 

Water Analysis Report

	neceived.		Lau No:	· Kaninghani I
	(		Y1178	G
]	Reported:	5/25	Reviewed by	ns
g/I				
	Plant Name	SCAR - N	Mayer_	
	AddressM	AOURTON, D	Jiac.	······
			TOF	
	Attention			
	Type of Sample	<u>GRAD</u>	II Sugar	10
TA	Sample Point	1-5-5	- Jun	1. PM
TA	Date of Collection Treatment Product(s)	Non P	Time _	- F Y 1
	PLANT			[
		WATER A		1
ig/l	рн	NH3	ITOC	Fe-Total
	A		Colloidal SrO2	Cu-Total
	MO	so,	Solvent Soluble	- Mn-Total
	в	- \$i02	Surfaciani	Al-Total
	Acidity	- NO,	CS-Water	Ni-Total
	Un-neutralized Conductivity	PO <sub>4</sub> -ortho	CS-Glycol	Zn-Total
	Neutralized Conductivity	-PO,-poly	Na EDTA-T	Ca-Total
	Dissolved	-PO-Organic	Na,NTA-T	Mg-Tolal
	Susp. Solids	Hardness-diss.	Na <sub>3</sub> NTA-Free	Hex CrO4
	Total Solids	Ca-diss.	EDTA-NTA Total	Total CrO.
	• Color	LMg-diss.	EDTA-NTA Free	Tri CrO,
	- Turbidity	-Na		
	CO2	- K FT	FLOURIDE	
	1512	TRACE A		·
	PO. (<3/19/1)	Fe (1-100 ppb)	Zn (5-100 ppb)	Na (1-500 ppb)
g/I	CI (.1-5 mg/i)	Cu (1-70 ppb)	Са (1-100 ррб)	AI (5-100 ppb)
<u>,</u>	SiO <sub>2</sub> (2-250 ppb)	Ni (1-150 ppb)	Мд (1-500 ррb)	TOC (25-5000 ppb
	Fuel Oll Analysis:	12 Oil 🗌 #6 Oil	0	
	Na	3 V	Ash	API Gravity
			<u></u>	BUSAU
	Engineer G	NEY LIFE	Olyge Code L	168
	521-Intern'i	- 525-Cooling	529-Sp. Accts	532-Consulting
	523-Oilfield	526-Polymer	531-Boiler Chem	590-Wholesaler
	$\perp D$	Date Results Needed:	57211	1781
li				Form 5060P
-				
<u>ig/I</u>			·····	

SUBSIDIARY OF MERCH & CO., INC PHYSICAL PROPERTIES OH at 25°C 7. 7 A reading mI N/30 H<sub>2</sub>SO<sub>4</sub>/100 mI CS in Water MO reading 20.7 CS in Glycol mI N/30 H<sub>2</sub>SO<sub>4</sub>/100 mI B reading mI N/30 H2SO4/100 mI Na<sub>4</sub>EDTA-Total **Total Acidity** as mg/I CaCO<sub>3</sub> Na<sub>3</sub>NTA-Total Free Mineral Acidity as mg/I CaCO<sub>3</sub> Na<sub>3</sub>NTA-Free Conductivity (un-neutralized) 910 Na₄EDTA/NTA-Total µmhos/cm as Na Conductivity (neutralized) Na<sub>4</sub>EDTA/NTA-Free µmhos/cm as Na<sub>3</sub> Dissolved Solids @ 180°C 500 Hydrazine mg/l Suspended Solids @ 105°C/07<5 mg/l Calsofix Total Solids @ 180°C 520 mg/l Specific Gravity METALS\* Color Platinum Cobalt Units < 5 1.0 2020 N.T.U. hron (Fe) :: bidity ample Description (visual): Waty Wilite. Copper (Cu) 20.05 1 sar Manganese (Mn) CHEMICAL ANALYSIS Aluminum (AI) mg/l Nickel (Ni) Hydroxide (OH) Zinc (Zn) Carbonate (CQ<sub>3</sub>) Calcium (Ca) Bicarbonate (HCO<sub>2</sub>) Magnesium (Mg) Hex Chromium as CrO<sub>4</sub>-dissolved Carbon Dioxide (CO<sub>2</sub>) 0.05 Total Chromium (CrO<sub>4</sub>) Ammonia (NH<sub>2</sub>) 42 Chloride (CI) Tri Chromium as CrO4-calc. Sulfate (SO<sub>4</sub>) 55 Lead (Pb) Silica (SiO<sub>2</sub>) 15 Barium (Ba) Nitrate (NO<sub>3</sub>) 15 Ortho Phosphate (PO,) TRACE ANALYSIS ×0.05 Polyphosphate (PO4)-calc. <0.05 Organic Phosphonate (PO<sub>4</sub>)-calc. <0.05 Ortho Phosphate (PO4) Fluoride (F) Polyphosphate (PO<sub>4</sub>)-calc. < 0.1 Organic Phosphonate (PO<sub>4</sub>)-calc. Trace Chloride (CI) Hardness as CaCO3-dissolved CAK. 384 Trace Silica (SiO<sub>2</sub>) Calcium (Ca)-dissolved Trace Iron (Fe) 75 Magnesium (Mg)-dissolved 42 Trace Copper (Cu) Sodium (Na) Trace Nickel (Ni) 18 Potassium (K) Trace Zinc (Zn) 1.8 (3) Trace Calcium (Ca) (1) (2) Trace Magnesium (Mg) (1) (2) (3) Total Organic Carbon (Non-Volatile) Trace Sodium (Na) (2)(3) (1) Colloidal (Non-reactive) SiO2 Trace Aluminum (AI) Oil/Grease Total Organic Carbon (TOC) Surfactant-MBAS Purgeable Organic Carbon (POC) µg/I

\*Unless stated otherwise, all metals are total (dissolved & particulate)

MADISON DEPARTMENT OF PUBLIC HEALTH LABORATORY REPORT

aboratory Number: ample: OSCA Collected by: Date Collected	N BENN	ETT			
Joneored Dy:	13 JUNE	1983			
Vantham			•		
Veather:					
Water Conditions:					
Results: (Form)	(Unit if not	ррмі			
Color	(Color Units)	والمستقد والشراك ومراق والمستقد ببهيهم	(Type)		
Jolor Odor	(Threshold O		(Type) (Type		
	(Turbidity Ur		·-160		
Sample D.O.	,- aronarcy of		· · · · · · · · · · · · · · · · · · ·		
5 Day B.O.D.					
C.O.D.					
Total Phosphorus			(P)		
Soluble Phosphorus	(Total)		(P)		
Soluble Phosphorus			(P)		
Total Solids					
Suspended Solids					
Dissolved Solids					515
Ammonia Nitrogen	(N)				······································
Organic Nitrogen	(N) ·				
Total Kjeldahl Nitr	· ·				
Nitrate Nitgrogen	(N)				
Nitrite Nitrogen	(N)				
Nitrite Nitrogen +					
(Total Nitrogen)	(N)		·····		
рН	(pH Units)				7.35
Alkalinity					330
Hardness			<b></b>		420
Chlorides	(C1)				880
Specific Conductan		/CM)			000
Sufactants	(ABS, LAS)				
Soluble Potassium	$(\mathbf{K})$		,		1.2
Soluble Sodium					20
Soluble Calcium	(Ca)				84
Soluble Magnesium	(Mg)				<u> </u>
Sulfate	$(So_4)$				
Sulfite	$(H_2S)$				
Sulfide	(203)	(np) \			
Inorganic Mercury	(Hg)	(PPb)			
Organic Mercury	(Hg)	(PPb)		Saluble	Arid Saluble
Total Mercury Silver	(hg)	(PPb)		Soluble • 00	Acid Soluble
	(ag)			• 00 • 0	
Barium	(Ba)			. 003	
<u>Cadmium</u> Chromium	(Cd)			100 3	
Chromium Cobalt	(Cr)				
	(Co) (Cu)			.01	
Copper Iron	(Cu) (Fe)		*****		
Iron Lead	(Fe) (Pb)			,01	
Manganese	(PD) (Mn)			,05	
Nickel	(Ni)			.03	
Tin	(NI) (Sn)				
Zine	(Sn) (Zn)			.04	
ومتله فيوميانه بميناه المربيا الأكري فالتكر ويعتمه ليوجه والمربية فواقدته والمراجعة المربية	and the second		••••••••••••••••••••••••••••••••••••••		
Total Heavy Meta				• [	
Arsenic	(As)				
Selenium	(Se)				

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ooratory Number:	-30-15		,
$ \begin{array}{c} \text{ople:} \\ 0 \\ \mathcal{M} \\ \end{array} $	WELL #3 ABE	ERG-	
lected by: B. SAL	ET		
	1HR BÍ		
ne Collecteã: 🛛 🤺	M		
Temperature:			
ther:			
er Temperature:			
cer Conditions:			
Results: (Form)	(Unit if not PPM)		
Color	(Color Units)	(Type)	······································
Odor	(Threshold Odor No.)	(Type)	
	(Turbidity Units)	(Type)	
Turbidity	(Turbially Units)		
Sample D.O.			
5 Day B.O.D.			
C.O.D.	وروان می از این از این		
Total Phosphorus		(P)	• 00
Soluble Phosphoru:		(P)	
Soluble Phosphorus	(Ortho)	(P)	• 00
Total Solids			
Suspended Solids			
Ammonia Nitrogen	(N)		
Organic Nitrogen	(N)		
Nitrate Nitrogen	(N)		1.1
Nitrite Nitrogen	(N)		
Nitrite Nitrogen +	Nitrate (N)		
(Total Nitrogen)	(N)		
pH	(pH Units)		
Alkalinity	(T 211200)		
Hardness			395
Chlorides	(C1)		42.2
Specific Conductan	•		870
Sufactants	(ABS,LAS)	•	-
Soluble Potassium	(K)		1,4
Soluble Sodium	(Na)		
Soluble Calcium			19
	(Ca)		78
Soluble Magnesium	(Mg)		45
Sulfate	(504)		
Sulfite	(H <sub>2</sub> S)		
Sulfide	(So <sub>3</sub> )		
Inorganic Mercury	(Hg) (PPb)		
Organic Mercury	(Hg) (PPb)		
Total Mercury	(hg) (PPb)		Acid Soluble
Silver	(ag)		.00 .00
Barium	(Ba)		.05 .05
Cadmium	(Cd)		.002 .001
Chromium	(Cr)		.00 .01
Cobalt	(Co)		.00
Copper	( <u>.</u> Cu)		•01 •01
Iron	(Fe)	·	
Lead	(Pb)		· · ·
Manganese	(Mn)		·01 ·02 ·05 ·04
Nickel	(Ni)		
Tin			.01 , .00
Zinc	(Sn) (Sn)		
	(Zn)		.01 .01
Total Heavy Metals			.2 .3
Arsenic	(As)		
Selenium	(Se)		

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UNIT WELL # 7		/-:•	/.\9	/、\9	/``\	/-^'	A. 0	$\langle \hat{e} \rangle$	N. M.		$\langle \hat{y} \rangle$	X° Ý			Y X	) OC
	1	v./	8: / L	3 { {	21/3	3.00			S X R	LYN N	- int	· A.			S.X.A	Ì,
	/	/ 4	1/ 5	1/ 1	?) °			J's	75° V	73	\$5		\$5's	\$8``s	93 X	S.
Alkalinity	326	318	318	328	328					2.4		3/2	330	340	327	
<u>Hardness</u> Total Solids	350	335	<u>351</u> 574	362	356	<u>347</u> 116344	<u>345</u>	346	338	340	<u></u>	322	375	322	397	
Turbidity	anapar gin	1000	in the second second		100	- 244	in de la compañía de		378-92 <sup>7</sup>				$\left\{ \left\{ e_{i}^{2},e_{i}^{2}\right\} \right\} =\left\{ e_{i}^{2},e_{i}^{2}\right\} =\left\{ e_{i}^{$	<u></u>	•7	
Specific Conductance		an a				Stand .	675	718	671	640	655	675	700	560	655	
Temperature		~ 왕(영구) - 111 - 111	स्तिम् । जनसम्बद्ध		and the second s						ur hagan Turingkan					<u> </u>
DISOLVED SOLIDS	7,2	7.3	7.5/7.2	7.6	7.9	344	1.15	11	7.36	7.2	•	7.28	7.39	7.34	7.46	-
Calcium 🔹	70	70	73	62	70	73	1112	69	64	68	62	72	70	75	78	
Chlorides	9.0	0.0	2	ス	2	5.2	3.5	2.1	2.5	3.9	3.8	4.2	AN ANT	3.5	3.5	
Eluorides	5.1	0.0	0.2	.20	<./	1.3			20		ાગમન્દ્રીનું કુ કુ			Algebra (1)	100	-
Magnesium .	42	35.5	41	46	44	40	jejst Mark Pres	39	39	38	<u> </u>	38	_4/	41	43	$\left  - \right $
Nitrogen - Total		and the second s				and a second s	nyesseri Græddin	riter (new) New (new)			a financial de la companya de la com La companya de la comp	- Manual Constant Second Constant Second Constant	an a	and the second se		<b> </b>
Ammonia - N		e kyzar	અન્યે સ્વધુન ન	-994-97 -	den.		(Activity)	inger sont		36940221) -	.04	.04	- AN DE SERVICE	000	.10	Γ
Organic - N		387 GP+	194 <sup>1</sup>	n des e		1999	Novigela		1262842		.04	Statistica (	sillingderseri 	er angeles		F
<u>Nitrate - N</u> Nitrite - N		0.0	5.5	,16	and the form	-/-		and and a second se	,75	.50	.25	.75	0.25	0.1	.35	$\vdash$
Nitrate + Nitrite - N		0.0	.003		1.5	jansi i							alite de las			+
TOT PIS. DEG. CARBON						4565		10.002								F
Phosphate - Total		station a	가지 못 있다.		a de gra	an an that a start a st The start a start	Ang a		:02	.00	.01	Stationess	And the second		NUMBER OF	0
S <b>olu</b> ble PO <sub>4</sub> - (Total)	$\sim 10^{-1}$		an an tari Tarih									$\begin{array}{c} & & & & & \\ & & & \\ & & & \\ & & & & & \\ & & & & \\ & & & & & \\ & & & & & \\ & & & & \\ & & & & & \\ & &$				1
Soluble PO4 - (Ortho.)					Alexandra Galary II.	aliga an Maria			.00	.00	.01	:01		Magazene		
I SSIUM	126					1.0	1.5	1.3	1.3	1.4	1.2	1.3	1.4	网络新闻	1.4	<u>,</u>
Soalum		- 41, <sup>1</sup> 819	3	4	4.1	4.4	4.3	3.8	3.9	4.1	3.6	3.9	3.9	4.4	4.0	+
<u>SeleMiun</u> Sulfate	11.4	14.5	22	24	29	23		21	-001	46	ng balan di ta Tanan di tanan di tan	.00	.000	1.002	.000	+-
Hydrogen Sulfide	11.4	1700	d d	-27_	-27-	03		34	77 1995-	70	in Action	4/	12		- 100 Miles - 100	+
Mercury - Inorganic							1.1					reditario - o	a station			+
Mercury - Organic	2.5		94 1		definite and the			a de la composition de la comp	ar daga sa		an de la casa de la cas La casa de la	- Service and	generalisa.		SEARS	Γ
Mercury - Total		gat 1	- 55		ļ	an ta a s		· · · · · · · · · · · · · · · · · · ·	0324(5)	ajata di T	i Kenya ya	.0001	19. al 19.		1.0005	1
ARSENIC Cadmium								;00	.00		,004	.005	.00	.000	:002	-
Chromium				1.11	<u> </u>	1		,00	.000	.600	.00	.00	.00	.00	,00~	+
Cobalt		1.1 1.						.0	.0	.0	.0	.0	.0	- de la dela dela	.0	
Copper	ļ	a start and a start and a start	1997) 	e de s		, ·		. //	.02	.00	.00	.01	.00	.00	.00	
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Lead SILVER		na an an Trachtan			<u> </u>		n a Araaga Araagaa	100		.00	.00	.00	.00	.00	.00	+
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carbonate								1.389.68						Sector Sector		
and the second secon			- <u>1</u>	1/22	2/36	10	2120		1			1 07	1.20	0.81	0.78	
Na:Cl Ca:Mg	1.19	1.50	1.08	1.00	2:32 1.07	1.37	1.2		1.30	1.16	1.22	. 87				
Cailly	1.17	1.4	1.08	1.23	1.07	0.44	1.08		1.09	1.14	1.13	1.19	1.11	1.3	1.10	
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UNIT WELL #	15	A Line	B		·		
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lardness 'otal Solids		341 365					
urbidity	367	560		· · · · · · · · · · · · · · · · · · ·		<u> </u>	
Specific Conductance	613	6/2					
emperature		6.1.					
-					-		
<u>H</u>	7.39	7.52					
Calcium	67	71	an Anna an Anna Anna Anna Anna Anna Ann				
Chlorides	9.5	9.9					
lagnesium	36	38					
идисотин 	36	100	<u></u>				
Vitrogen - Total	1						
mmonia - N	.04	.01			<u> </u>		
rganic - N							
litrate - N	1.9	1.8					
litrite - N	ļ						
Nitrate + Nitrite - N	ļ						
hosphate - Total	+						
Soluble PO <sub>4</sub> - (Total)	+						
sol le $PO_4$ - (Ortho.)							
otassium	1.3	1.4					
odium	5.0	5.2					
SELENIOM	.000	.000					
Sulfate		21.8					
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RARIUM	.02	.02					
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Fre CO <sub>2</sub>							
Carbonate							
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Naicl	0.82	0-21 1.14					
Gaine	1.10	1 12/					
Callie	1010	1.17					

